



HAL
open science

CO₂-Dissolved and Aqueous Gas Separation

G. Blount, M. Gorenssek, L. Hamm, K. O'neil, Christophe Kervévan

► **To cite this version:**

G. Blount, M. Gorenssek, L. Hamm, K. O'neil, Christophe Kervévan. CO₂-Dissolved and Aqueous Gas Separation. Energy Procedia, Elsevier, 2016, 114, pp.2675 - 2681. 10.1016/j.egypro.2017.03.1451 . hal-03702853

HAL Id: hal-03702853

<https://hal-brgm.archives-ouvertes.fr/hal-03702853>

Submitted on 23 Jun 2022

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



13th International Conference on Greenhouse Gas Control Technologies, GHGT-13, 14-18
November 2016, Lausanne, Switzerland

CO₂-Dissolved and Aqueous Gas Separation

G. Blount^a ; M. Gorenssek Ph.D., P.E.^a; L. Hamm Ph.D.^a; K. O'Neil^b;

C. Kervévan, Ph.D.^c

^aSavannah River Site, and Savannah River National Laboratory, Aiken, SC 29808, USA

^bPartnering in Innovation, Inc., Orlando FL, USA; ^cBRGM, 3 av. Claude Guillemin, BP 36009, 45060 Orléans Cedex 2, France

Abstract

Partnering in Innovation, Inc. (Pi-Innovation) offers a novel aqueous gas separation and carbon capture approach. Specific features in the design uniquely integrate multiple energy recovery processes and the potential for contaminant removal (i.e., oxides of sulphur (SO_x), nitrogen (NO_x), selenium (SeO_x), and mercury (HgO_x). CO₂-DISSOLVED is an ANR multi-national project led by the BRGM, with Phase II funded as one of the first Geodenergies projects. Phase I explored the techno-economic feasibility of geothermal energy production combined with CO₂ injection using a simplified version of the Pi-Innovation technology. Here we review the Pi-CO₂capture process and the integration with dissolved injection applications.

© 2017 The Authors. Published by Elsevier Ltd. This is an open access article under the CC BY-NC-ND license (<http://creativecommons.org/licenses/by-nc-nd/4.0/>).

Peer-review under responsibility of the organizing committee of GHGT-13.

Keywords: Carbon Capture; Enhanced Oil Recovery; EOR; Greenhouse Gas; GHG; Geothermal Energy; Dissolved CO₂; Post-Combustion

1. Introduction

CO₂ Dissolved and Partnering in Innovation Inc. (Pi-Innovation) are collaborating to integrate aqueous CO₂ storage with aqueous gas separation. Aqueous capture and storage of CO₂ storage has applications including co-injection with reservoir flooding, with reproduced water in oil production, and in geothermal circulation systems. Pi-Innovation offers a patented aqueous gas separation and carbon capture technology that is highly promising in terms of potential total system cost reduction and energy recovery.

The CO₂-Dissolved project evaluated the storage potential and techno-economic feasibility of co-injecting dissolved CO₂ in a geothermal doublet in the Paris Basin.

2. CO₂-DISSOLVED

CO₂-DISSOLVED is a multi-national project led by the Bureau de Recherches Géologiques et Minières (BRGM) and funded by the French National Research Agency (ANR) with Phase II funded as one of the first Geodenergies projects. Geodenergies is a French industry-driven initiative grouping 18 companies and research organizations aiming at: (1) structuring a community of expertise to promote subsurface energy technologies that are key to a global energy transition; (2) cross-fertilizing to develop 3 emerging industrial sectors: geothermal energy production, CO₂ storage, and energy storage.

As the name “CO₂-DISSOLVED” implies, the project focuses on the feasibility of the injection of dissolved CO₂ in water, as opposed to supercritical CO₂ injection. In this, and drawing from a process originally developed at a US Department of Energy (DOE) laboratory, Partnering in Innovation, Inc. (Pi-Innovation) offers a novel aqueous gas separation and carbon capture approach.

The CO₂-Dissolved concept explores the storage of CO₂ dissolved in water rather than in supercritical form; Kervévan et al. [1] and Kervévan et al. [2]. Figure 1 illustrates the carbonation of spent geothermal water in an injection well associated with a geothermal doublet Kervévan [3]. The dissolved form of CO₂ storage in a saline aquifer is neutral to negatively buoyant, and shifts the geologic storage mechanism to the solubility trapping realm. Supercritical trapping relies on structural and stratigraphic trapping, and residual trapping, which is positively buoyant. Positive buoyancy represents a greater risk of uncontrolled release of CO₂ to the surface. Figure 2 is a modification of the typical IPCC [4] geologic storage effectiveness diagram illustrating the range of CO₂ trapping mechanisms versus storage security. Moving directly to solubility trapping offers lower brine displacement in the vicinity of the injection and significantly reduces the time required to achieve the highest level of geologic storage security.

The storage potential for CO₂ in geothermal brine is limited by the solubility of the CO₂ in the brine and the purity of the CO₂ that is to be dissolved in the brine; due to gas competition in the dissolution process. This project targets low to medium range CO₂ emitters (~10 to 150 kilotons/year) that could be compatible with a single doublet geothermal installation adjacent to the facility.

The coupling of geothermal heat with CO₂ storage adds the potential for energy use and revenue generation, and adds the opportunity for valorization of the injection operations. In France, 650 potentially compatible industrial sites have been identified. These sites represent 25% of the stationary emissions for the country Kervévan [3].

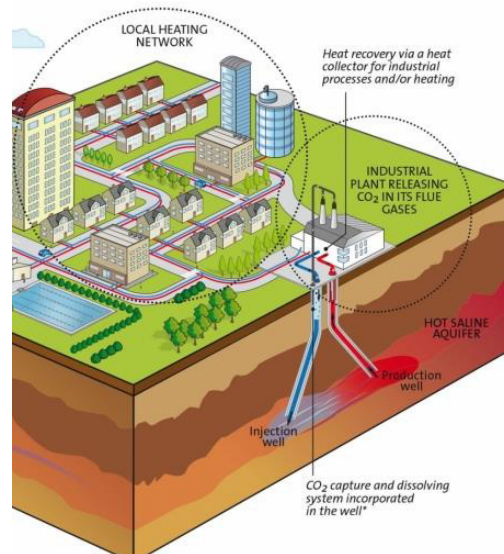


Figure 1 CO₂-Dissolved: geothermal doublet injection well; Modified from BRGM, Kervévan [3]

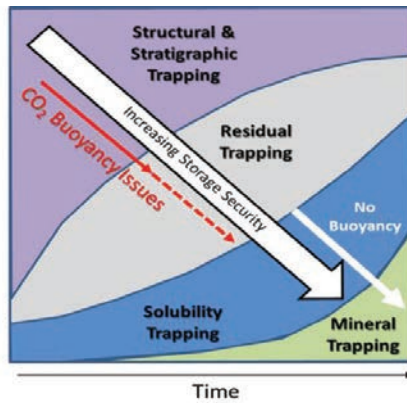


Figure 2 CO2 Trapping Mechanisms and Relative Storage Security Over Time

3. Pi-CO2

Partnering in Innovation Inc. offers post combustion CO₂ capture using water as the physical mass transfer solvent. The patented Pi-CO₂ process can be deployed on land or offshore, and has the capacity to remove SO_x, NO_x, and vaporized metals from the flue gas stream along with the CO₂. The relatively complete treatment of flue gases adds the potential for substantial cost savings, providing capture incentives to a wide variety of flue gas sources typically containing moderately high CO₂ contents (i.e., coal, cement, burnt lime, refining, smelting, biomass combustion, incineration, etc.).

The CO₂ capture component of the process takes advantage of the solubility difference between CO₂ and N₂ in water at higher pressures and lower temperatures. Figure 3a illustrates the saturation concentrations of CO₂ and N₂ in water at 30 C, versus pressure. Figure 3b illustrates the saturation concentration of CO₂ at various temperatures and pressures.

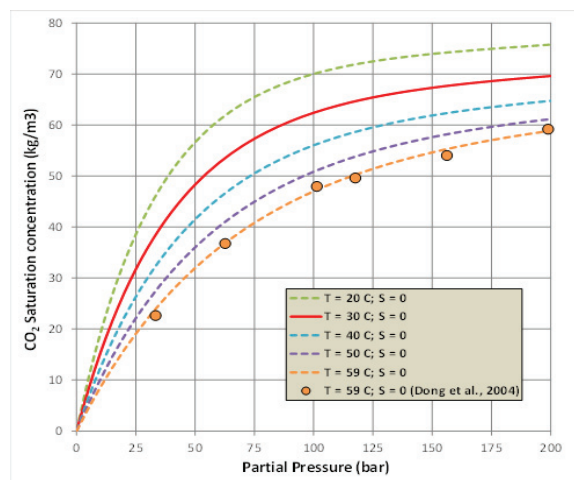
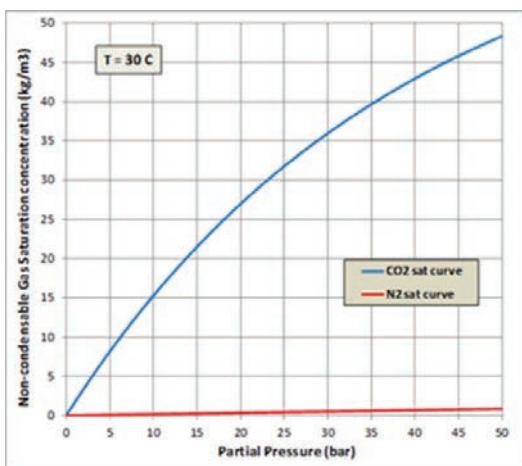


Figure 3a CO₂ and N₂ Saturation Concentration versus partial pressure at 30 C dissolved in pure water and 3b CO₂ Solubility in pure water across a range of temperature and partial pressure conditions

Capture of CO₂ occurs as water is circulated through a closed loop absorber and pressure swing desorber; the water in the closed loop is used “over and over again”. High hydrostatic pressure magnifies the CO₂ carrying capacity of the water. Lower temperature locations also further increased the CO₂ carrying capacity. High hydrostatic pressure is achieved by placing the absorber/desorber in a water filled large diameter shaft, or through submergence in an offshore setting. Lower water temperatures can be acquired through submergence in relatively cold ocean or lake settings or by heat exchange with a natural cold water source.

The configuration of the absorber/desorber system is down-flow for the absorber and up-flow for the desorber. The absorber contains many stages that support sequential loading of CO₂ into the water. Figure 4 shows the transfer of CO₂ from the gas phase to the liquid phase in a down-flow absorber having 16 stages of mass transfer, operating at 30 C, with approximately 14% CO₂ in the flue gas (similar to coal). The absorber will be more efficient with higher CO₂ concentration emissions (i.e., cement, smelting and refining cat-crack), and less efficient at lower CO₂ concentrations (i.e., natural gas fired turbines, natural gas fired power plants).

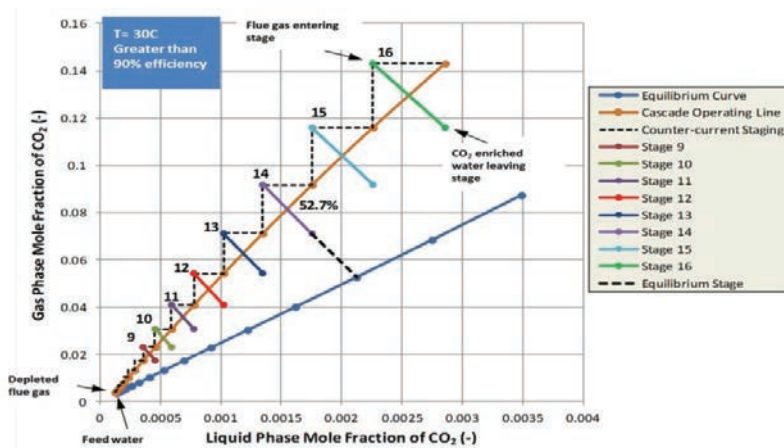


Figure 4 Step-wise CO₂ transfer from the gas-phase to water-phase with a multi-stage down-flow cascading absorber.

The absorber efficiency will increase with: higher hydrostatic pressure, lower temperature, optimum bubble fraction, with longer absorber stages, and with an increase in the number of stages. The presence of free and dissolved N₂ in the water in the absorber acts as a poison with respect to clathrate formation at the lowest conceivable temperature and highest conceivable pressure that the system is likely to be operated.

The desorber is configured to support sequential desorption promoting higher CO₂ concentration in the final CO₂ product and supporting gas lift pumping within the desorber. Gas lift pumping within the desorber drives the circulation within the absorber/desorber loop because the upward flow in the desorber achieves down-flow in the absorber.

Gas lift pumping is a well understood technology. However, most gas lift pumps are not typically on the scale of the Pi-CO₂ process (diameter of the desorber riser and depth of submergence), and are not generally driven by effervescing gases (greater efficiency). Pump efficiency is controlled by the diameter of the rising pipe, bubble dispersion, gas volume, submergence ratio, and the height in which the pumped water is lifted; Awari et al. [5], Kassab et al. [6], and Barrut et al. [7]. The best performance in volume of water pumped can be achieved with: the largest diameter riser pipe, a high submergence ratio (submergence/lift height), an optimum volume of gas to make bubbles, and good bubble dispersion.

Fan et al. [8] performed a study of gas lift pumping with a relatively large diameter rising pipe, and relatively low water lift. Pi-Innovation developed a simple model using the empirical data derived from the study. A comparison with the modeling work of Fan et al. [8] suggests that there is more than sufficient gas to support pumping in the Pi-CO₂ process. The challenge may be to selectively remove gas to maintain the optimum gas volume for optimum pumping in the desorber.

An important part of the Pi-CO₂ process is the energy recovery from the pressurized non-dissolved gas that exits the top of the absorber. The process uses turbo-machinery to recover a large fraction of the compression energy penalty by combining the heat of compression with the pressurized gas. The Pi-CO₂ process has similar parasitic energy requirements when compared to competing technologies (~22%) in a coal fired setting, but can take advantage of system waste heat for electricity production that offsets CO₂ to an equivalency of 16 to 18%. The process can be driven by direct natural gas firing with alternate fuels, or driven with electricity. Direct firing can offer greater energy efficiency. The turbo-line can be electrically linked or mechanically linked as in some Compressed Air Energy Storage turbo-lines. Figure 5 shows a generalized sketch of the Pi-CO₂ system.

Uniquely, the Pi-CO₂ process removes SO_x, NO_x, vaporized metals, and CO₂, in a single integrated process. The oxides are removed in compression condensate and at inter-cooler and after-cooler steps during flue gas compression. Figure 6 shows a simple chemical model output based upon Oxyfuel combustion flue gas compression empirical data, sourced from White et al. [9] and Huckaby [10].

SO_x and NO_x removal is usually required prior to CO₂ capture because the presence of these compounds accelerates the degradation of typical capture chemicals (amines, sorbents, etc.); Rezaei and Jones [11], Supap et al. [12], Rohr et al. [13], and El Moudir [14]. Based upon CAPEX and OPEX costs from Cichanowitz [15], SO_x pretreatment, alone, prior to CO₂ capture, can add as much as \$25 to \$30 (US Dollars) to the cost of a metric ton of CO₂ removed.

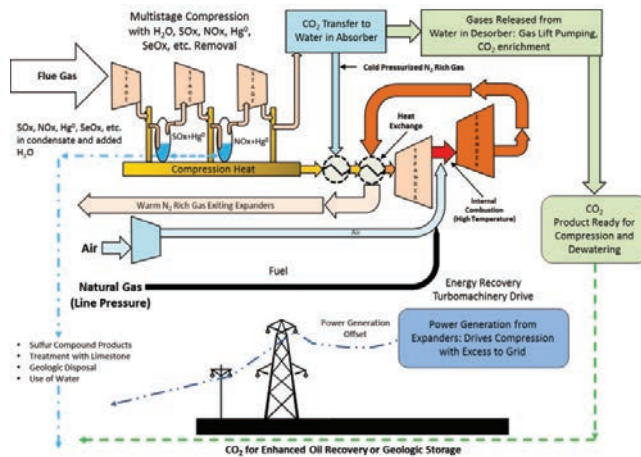


Figure 5 Generalized Pi-CO₂ capture processes showing general energy recovery and flue gas contaminant removal

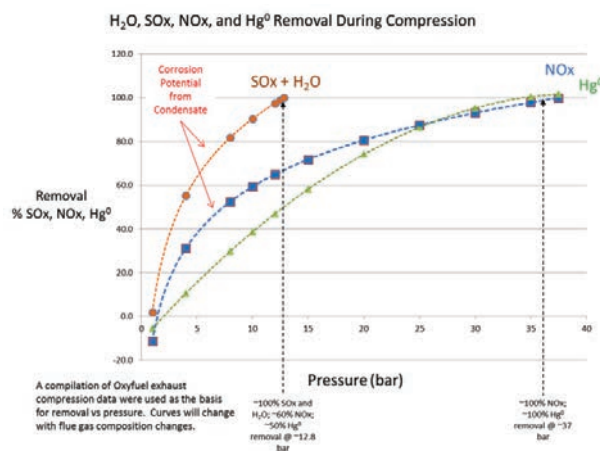


Figure 6 Simple chemical model output based upon empirical data from Oxyfuel flue gas compression studies.

The addition of heat to the turbo-line can allow for excess power generation to offset capital cost. Turbo-machinery is a very large part of the capital cost of the process. The potential to generate electricity is also useful because it can offset a portion of higher carbon sourced power on the grid in some locations which can improve the overall carbon reduction efficiency. Figure 7 illustrates this overall carbon reduction with electricity offset.

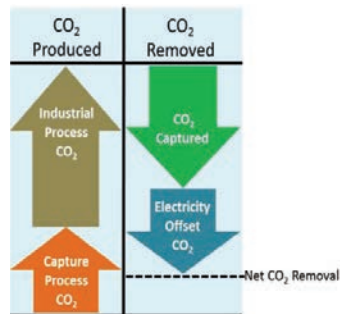


Figure 7 Overall carbon reduction with electricity offset

Energy demand is typically considered an important measure of operating cost between competing CO₂ capture technologies. The Pi-CO₂ process efficiently uses the power required for operation by recovering a large fraction of the compression energy penalty, using the energy of compression of the dissolved gas fraction to drive absorber/desorber pumping, and provides an opportunity to generate electricity offsets with the turbo-line.

CO₂ capture with Pi-CO₂ uses no expensive chemicals, has no significant chemical degradation, and no chemical wastes. The overall footprint of the technology is less because the absorber/desorber system is deployed in the subsurface or subsea. The capital cost can be lower when pretreatment for SO_x, and NO_x are considered, and future capital costs can be lower when Hg₀ removal is considered. Operational costs are also low due to the use of long life turbomachinery for compression and energy recovery rather than chemical intensive processes.

4. Integration of CO₂-DISSOLVED and Pi-CO₂

The integration of aqueous CO₂ storage with aqueous gas separation can offer lower risk geologic storage scenarios in situations where water is available for injection into saline aquifers. It is possible to capture CO₂ from various industrial sources and produce a high purity CO₂ product for use as supercritical CO₂, or use relatively pure CO₂ dissolved in water with the Pi-CO₂ absorber technology, and directly inject the water into the subsurface. It is also possible to capture the CO₂ from flue gas with the Pi-CO₂ process absorber and inject the absorber discharge into the subsurface.

References

- [1] Kervévan C., Beddelem M.-H., Blount G., O'Neil K. CO₂-DISSOLVED: a novel concept coupling geological storage of dissolved CO₂ and geothermal heat recovery – Part 1: Assessment of the integration of an innovative low-cost, water-based CO₂ capture technology. *Energy Procedia, GHGT 12 Conference Proceedings*, Austin, TX, USA; Oct. 5-9, 2014.
- [2] Kervévan C., Bugarel F., Galiègue X., Le Gallo Y., May F., O'Neil K., Sterpenich J. CO₂-DISSOLVED: A Novel Approach to Combining CCS and Geothermal Heat Recovery. In: EAGE (Ed.), *Sustainable Earth Sciences, Technologies for Sustainable Use of the Deep Sub-surface*, Pau, France, Sept. 30 - Oct. 4, 2013.
- [3] Kervévan C. Successfully combining CO₂ storage and geothermal heat extraction. *BRGM Flag Ship Project Web Article*, <http://co2-dissolved.brgm.fr> 2016.
- [4] IPCC. *IPCC Special Report on Carbon Dioxide Capture and Storage*. Prepared by Working Group III of the Intergovernmental Panel on Climate Change [Metz, B., O. Davidson, H. C. de Coninck, M. Loos, and L. A. Meyer (eds.)]. Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 2005, 442 pp.
- [5] Awari, G.K., Ardhapurkar, P.M., Wakde, D.G., Bhuyar, L.B., 2004. Performance analysis of air-lift pump design, *Proc.IMECH E Part C: J. Mech. Eng. Sci.* 218; 2004, 1155–1161.

- [6] Kassab, S.Z., Kandil, H.A., Warda, H.A., Ahmed, W.H., 2009. Air-lift pumps characteristics under two-phase flow conditions, *Int. J. Heat Fluid Flow* 30; 2009, 88–98.
- [7] Bertrand Barrut, Jean Paul Blancheton, Jean Yves Champagne, and Alain Grasmick, 2012. Mass Transfer efficiency of a vacuum airlift- Application to water recycling in aquaculture systems. *Aquacultural Engineering*, Vol 46, January 2012, 18-26.
- [8] W. Fan, J. Chen, Y. Pan, H. Huang, C. Chen, Y. Chen. Experimental study on the performance of an air-lift pump for artificial upwelling. *Ocean Engineering*, 59; 2013, pp 47-57.
- [9] V. White, R.J. Allam. Purification of oxyfuel-derived CO₂ for sequestration or EOR, *Proceedings of the 8th international conference on greenhouse gas control technologies*, Trondheim, Norway; 2006, 19–22.
- [10] E. D. Huckaby. U.S. DOE/NETL Office of Research and Development – Oxyfuel Combustion Progress Update Thermofluids; Slide Presentation. *Annual NETL CO₂ Capture Technology for Existing Plants R&D Meeting*, Pittsburgh, PA, March 24-26, 2009
- [11] Fateme Rezaei, and Christopher W. Jones. Stability of Supported Amine Adsorbents to SO₂ and NO_x in Postcombustion CO₂ Capture. 1. Single-Component Adsorption. dx.doi.org/10.1021/ie4019116 | *Ind. Eng. Chem. Res.* 52; 2013, 12192–12201
- [12] Teeradet Supap, Raphael Idem, Paitoon Tontiwachwuthikul, and Chintana Saiwan. Kinetics of sulfur dioxide and oxygen induced degradation of aqueous monoethanolamine solution during CO₂ absorption from power plant flue gas streams. *International Journal of Greenhouse Gas Control* 3; 2009, 133-142.
- [13] Annette Rohr, Jacob D. McDouals, Dean Kracko, Melanie Doyle-Eisele, Styephanie L. Shaw, Eladio M. Knipping. Potential toxicological effects of amines used for carbon capture and storage and their degradation products. *Elsevier Energy Procedia* 37; 2013, 759-768.
- [14] Walid El Moudir. *Experimental Studies for Development of a Purification Process for Single and Mixed Amine Solvents*, PhD Thesis in Industrial Engineering, University of Regina, Saskatchewan, 2012.
- [15] J. Edward Cichanowicz. Current Capital Cost and Cost Effectiveness of Power Plant Emissions Control Technologies. *Utility Air Regulatory Group, American Public Power Association*, www.publicpower.org, 2010.